



**Formulas
Examples
with Units**

List of 33 Important Disposing of the Sewage Effluents Formulas

1) Actual Dissolved Oxygen Formula ↻

Formula

$$A_{DO} = S_{DO} - D$$

Example with Units

$$4.8 \text{ mg/L} = 9 \text{ mg/L} - 4.2 \text{ mg/L}$$

Evaluate Formula ↻

2) Mixing Concentration Formula ↻

Formula

$$C = \frac{C_s \cdot Q_s + C_R \cdot Q_{\text{stream}}}{Q_s + Q_{\text{stream}}}$$

Example with Units

$$1.2 = \frac{0.2 \cdot 10 \text{ m}^3/\text{s} + 1.3 \cdot 100 \text{ m}^3/\text{s}}{10 \text{ m}^3/\text{s} + 100 \text{ m}^3/\text{s}}$$

Evaluate Formula ↻

3) River Stream Concentration Formula ↻

Formula

$$C_R = \frac{C \cdot (Q_s + Q_{\text{stream}}) - (C_s \cdot Q_s)}{Q_{\text{stream}}}$$

Example with Units

$$1.3 = \frac{1.2 \cdot (10 \text{ m}^3/\text{s} + 100 \text{ m}^3/\text{s}) - (0.2 \cdot 10 \text{ m}^3/\text{s})}{100 \text{ m}^3/\text{s}}$$

Evaluate Formula ↻

4) River Stream Flow Rate Formula ↻

Formula

$$Q_{\text{stream}} = \frac{(C_s \cdot Q_s) - (C \cdot Q_s)}{C - C_R}$$

Example with Units

$$100 \text{ m}^3/\text{s} = \frac{(0.2 \cdot 10 \text{ m}^3/\text{s}) - (1.2 \cdot 10 \text{ m}^3/\text{s})}{1.2 - 1.3}$$

Evaluate Formula ↻

5) Saturated Dissolved Oxygen Formula ↻

Formula

$$S_{DO} = D + A_{DO}$$

Example with Units

$$9 \text{ mg/L} = 4.2 \text{ mg/L} + 4.8 \text{ mg/L}$$

Evaluate Formula ↻

6) Sewage Concentration Formula ↻

Formula

$$C_s = \frac{C \cdot (Q_s + Q_{\text{stream}}) - (C_R \cdot Q_{\text{stream}})}{Q_s}$$

Example with Units

$$0.2 = \frac{1.2 \cdot (10 \text{ m}^3/\text{s} + 100 \text{ m}^3/\text{s}) - (1.3 \cdot 100 \text{ m}^3/\text{s})}{10 \text{ m}^3/\text{s}}$$

Evaluate Formula ↻

7) Sewage Flow Rate Formula ↻

Formula

$$Q_s = \frac{(C_R - C) \cdot Q_{\text{stream}}}{C - C_s}$$

Example with Units

$$10 \text{ m}^3/\text{s} = \frac{(1.3 - 1.2) \cdot 100 \text{ m}^3/\text{s}}{1.2 - 0.2}$$

Evaluate Formula ↻



8) Critical Oxygen Deficit Formulas

8.1) Critical Oxygen Deficit Formula

Formula

$$D_c = K_D \cdot L_t \cdot \frac{10^{-K_D \cdot t_c}}{K_R}$$

Example with Units

$$0.0002 = 0.23 \text{ d}^{-1} \cdot 0.21 \text{ mg/L} \cdot \frac{10^{-0.23 \text{ d}^{-1} \cdot 0.5 \text{ d}}}{0.22 \text{ d}^{-1}}$$

Evaluate Formula

8.2) Critical Oxygen Deficit given Self Purification Constant Formula

Formula

$$D_c = L_t \cdot \frac{10^{-K_D \cdot t_c}}{f}$$

Example with Units

$$0.0002 = 0.21 \text{ mg/L} \cdot \frac{10^{-0.23 \text{ d}^{-1} \cdot 0.5 \text{ d}}}{0.9}$$

Evaluate Formula

8.3) Critical Oxygen Deficit in First Stage Equation Formula

Formula

$$D_c = \frac{\left(\frac{L_t}{f}\right)^f}{1 - (f - 1) \cdot D_o}$$

Example with Units

$$0.0005 = \frac{\left(\frac{0.21 \text{ mg/L}}{0.9}\right)^{0.9}}{1 - (0.9 - 1) \cdot 7.2 \text{ mg/L}}$$

Evaluate Formula

9) Critical Time Formulas

9.1) Critical Time Formula

Formula

$$t_c = \left(\frac{1}{K_R - K_D}\right) \cdot \log_{10} \left(\left(\frac{K_D \cdot L_t - K_R \cdot D_o + K_D \cdot D_o}{K_D} \cdot L_t \right) \cdot \left(\frac{K_R}{K_D} \right) \right)$$

Example with Units

$$697.8548 \text{ d} = \left(\frac{1}{0.22 \text{ d}^{-1} - 0.23 \text{ d}^{-1}}\right) \cdot \log_{10} \left(\left(\frac{0.23 \text{ d}^{-1} \cdot 0.21 \text{ mg/L} - 0.22 \text{ d}^{-1} \cdot 7.2 \text{ mg/L} + 0.23 \text{ d}^{-1} \cdot 7.2 \text{ mg/L}}{0.23 \text{ d}^{-1}} \cdot 0.21 \text{ mg/L} \right) \cdot \left(\frac{0.22 \text{ d}^{-1}}{0.23 \text{ d}^{-1}} \right) \right)$$

Evaluate Formula

9.2) Critical Time given Self Purification Constant with Critical Oxygen Deficit Formula

Formula

$$t_c = \log_{10} \frac{D_c \cdot f}{K_D \cdot L_t}$$

Example with Units

$$0.4745 \text{ d} = \log_{10} \frac{0.0003 \cdot 0.9}{0.21 \text{ mg/L} \cdot 0.23 \text{ d}^{-1}}$$

Evaluate Formula

9.3) Critical Time given Self Purification Factor Formula

Formula

$$t_c = \left(\log_{10} \frac{1 - (f - 1) \cdot \left(\frac{D_c}{L_t}\right) \cdot f}{K_D \cdot (f - 1)} \right)$$

Example with Units

$$2.2839 \text{ d} = \left(\log_{10} \frac{1 - (0.9 - 1) \cdot \left(\frac{0.0003}{0.21 \text{ mg/L}}\right) \cdot 0.9}{0.23 \text{ d}^{-1} \cdot (0.9 - 1)} \right)$$

Evaluate Formula

9.4) Critical Time when we have Critical Oxygen Deficit Formula

Formula

$$t_c = \log_{10} \frac{D_c \cdot K_R}{K_D \cdot L_t}$$

Example with Units

$$0.5896 \text{ d} = \log_{10} \frac{0.0003 \cdot 0.22 \text{ d}^{-1}}{0.23 \text{ d}^{-1} \cdot 0.21 \text{ mg/L}}$$

Evaluate Formula



10) Deoxygenation Coefficient Formulas ↻

10.1) Deoxygenation Coefficient given Self Purification Constant Formula ↻

Formula	Example with Units
$K_D = \frac{K_R}{f}$	$0.2444 \text{ d}^{-1} = \frac{0.22 \text{ d}^{-1}}{0.9}$

Evaluate Formula ↻

10.2) Deoxygenation Constant given Self Purification Constant with Critical Oxygen Deficit Formula ↻

Formula	Example with Units
$K_D = \log_{10} \frac{D_c \cdot \frac{f}{L_t}}{t_c}$	$0.2183 \text{ d}^{-1} = \log_{10} \frac{0.0003 \cdot \frac{0.9}{0.21 \text{ mg/L}}}{0.5 \text{ d}}$

Evaluate Formula ↻

11) Oxygen Deficit Formulas ↻

11.1) DO Deficit using Streeter-Phelps Equation Formula ↻

Formula
$D = \left(K_D \cdot \frac{L}{K_R \cdot K_D} \right) \cdot \left(10^{-K_D \cdot t} \cdot 10^{-K_R \cdot t} + D_0 \cdot 10^{-K_R \cdot t} \right)$

Evaluate Formula ↻

Example with Units
$5.3649 \text{ mg/L} = \left(0.23 \text{ d}^{-1} \cdot \frac{40 \text{ mg/L}}{0.22 \text{ d}^{-1} - 0.23 \text{ d}^{-1}} \right) \cdot \left(10^{-0.23 \text{ d}^{-1} \cdot 6 \text{ d}} - 10^{-0.22 \text{ d}^{-1} \cdot 6 \text{ d}} + 7.2 \text{ mg/L} \cdot 10^{-0.22 \text{ d}^{-1} \cdot 6 \text{ d}} \right)$

11.2) Log value of Critical Oxygen Deficit Formula ↻

Formula	Example with Units
$D_c = 10^{\log_{10} \left(\frac{L_t}{T} \right) - (K_D \cdot t_c)}$	$0.0002 = 10^{\log_{10} \left(\frac{0.21 \text{ mg/L}}{0.9} \right) - (0.23 \text{ d}^{-1} \cdot 0.5 \text{ d})}$

Evaluate Formula ↻

11.3) Oxygen Deficit Formula ↻

Formula	Example with Units
$D = S_{DO} - A_{DO}$	$4.2 \text{ mg/L} = 9 \text{ mg/L} - 4.8 \text{ mg/L}$

Evaluate Formula ↻

11.4) Oxygen Deficit given Critical Time in Self Purification Factor Formula ↻

Formula	Example with Units
$D_c = \left(\frac{L_t}{f - 1} \right) \cdot \left(1 - \left(\frac{10^{t_c \cdot K_D \cdot (f - 1)}}{f} \right) \right)$	$0.0002 = \left(\frac{0.21 \text{ mg/L}}{0.9 - 1} \right) \cdot \left(1 - \left(\frac{10^{0.5 \text{ d} \cdot 0.23 \text{ d}^{-1} \cdot (0.9 - 1)}}{0.9} \right) \right)$

Evaluate Formula ↻

12) Oxygen Equivalent Formulas ↻

12.1) Oxygen Equivalent given Critical Oxygen Deficit Formula ↻

Formula	Example with Units
$L_t = D_c \cdot \frac{K_R}{K_D \cdot 10^{-K_D \cdot t_c}}$	$0.374 \text{ mg/L} = 0.0003 \cdot \frac{0.22 \text{ d}^{-1}}{0.23 \text{ d}^{-1} \cdot 10^{-0.23 \text{ d}^{-1} \cdot 0.5 \text{ d}}}$

Evaluate Formula ↻



12.2) Oxygen Equivalent given Critical Time in Self Purification Factor Formula ↻

Formula

$$L_t = D_c \cdot \frac{f - 1}{1 - \left(\frac{10^{t_c \cdot K_D \cdot (f-1)}}{f} \right)}$$

Example with Units

$$0.3655 \text{ mg/L} = 0.0003 \cdot \frac{0.9 - 1}{1 - \left(\frac{10^{0.5 \text{d} \cdot 0.23 \text{d}^{-1} \cdot (0.9 - 1)}}{0.9} \right)}$$

Evaluate Formula ↻

12.3) Oxygen Equivalent given Log value of Critical Oxygen Deficit Formula ↻

Formula

$$L_t = f \cdot 10^{\log_{10}(D_c) + (K_D \cdot t_c)}$$

Example with Units

$$0.3519 \text{ mg/L} = 0.9 \cdot 10^{\log_{10}(0.0003) + (0.23 \text{d}^{-1} \cdot 0.5 \text{d})}$$

Evaluate Formula ↻

12.4) Oxygen Equivalent given Self Purification Constant with Critical Oxygen Deficit Formula ↻

Formula

$$L_t = D_c \cdot \frac{f}{10^{-K_D \cdot t_c}}$$

Example with Units

$$0.3519 \text{ mg/L} = 0.0003 \cdot \frac{0.9}{10^{-0.23 \text{d}^{-1} \cdot 0.5 \text{d}}}$$

Evaluate Formula ↻

13) Reoxygenation Coefficient Formulas ↻

13.1) Reoxygenation Coefficient at 20 Degree Celsius Formula ↻

Formula

$$K_{R(20)} = \frac{K_R}{(1.016)^{T-20}}$$

Example with Units

$$0.22 \text{ d}^{-1} = \frac{0.22 \text{ d}^{-1}}{(1.016)^{20 \text{K} - 20}}$$

Evaluate Formula ↻

13.2) Reoxygenation Coefficient given Critical Oxygen Deficit Formula ↻

Formula

$$K_R = K_D \cdot L_t \cdot \frac{10^{-K_D \cdot t_c}}{D_c}$$

Example with Units

$$0.1235 \text{ d}^{-1} = 0.23 \text{ d}^{-1} \cdot 0.21 \text{ mg/L} \cdot \frac{10^{-0.23 \text{d}^{-1} \cdot 0.5 \text{d}}}{0.0003}$$

Evaluate Formula ↻

13.3) Reoxygenation Coefficient given Self Purification Constant Formula ↻

Formula

$$K_R = K_D \cdot f$$

Example with Units

$$0.207 \text{ d}^{-1} = 0.23 \text{ d}^{-1} \cdot 0.9$$

Evaluate Formula ↻

13.4) Reoxygenation Coefficients Formula ↻

Formula

$$K_R = K_{R(20)} \cdot (1.016)^{T-20}$$

Example with Units

$$0.65 \text{ d}^{-1} = 0.65 \text{ d}^{-1} \cdot (1.016)^{20 \text{K} - 20}$$

Evaluate Formula ↻

13.5) Stream Depth given Reoxygenation Coefficient Formula ↻

Formula

$$d = \left(3.9 \cdot \frac{\sqrt{v}}{k} \right)^{1.5}$$

Example with Units

$$42.2505 \text{ m} = \left(3.9 \cdot \frac{\sqrt{60 \text{ m/s}}}{0.11 \text{ s}^{-1}} \right)^{1.5}$$

Evaluate Formula ↻



13.6) Temperature given Reoxygenation Coefficient at T degree Celsius Formula

Formula

$$T = \log \left(\left(\frac{K_R}{K_{R(20)}} \right), 1.016 \right) + 20$$

Example with Units

$$19.9853^\circ\text{C} = \log \left(\left(\frac{0.22 \text{ d}^{-1}}{0.65 \text{ d}^{-1}} \right), 1.016 \right) + 20$$

Evaluate Formula 

14) Self Purification Constant Formulas

14.1) Self Purification Constant Formula

Formula

$$f = \frac{K_R}{K_D}$$

Example with Units

$$0.9565 = \frac{0.22 \text{ d}^{-1}}{0.23 \text{ d}^{-1}}$$

Evaluate Formula 

14.2) Self Purification Constant given Critical Oxygen Deficit Formula

Formula

$$f = L_t \cdot \frac{10^{-K_D \cdot t_c}}{D_c}$$

Example with Units

$$0.5372 = 0.21 \text{ mg/L} \cdot \frac{10^{-0.23 \text{ d}^{-1} \cdot 0.5 \text{ d}}}{0.0003}$$

Evaluate Formula 

14.3) Self Purification Constant given Log value of Critical Oxygen Deficit Formula

Formula

$$f = \frac{L_t}{10^{\log_{10}(D_c) + (K_D \cdot t_c)}}$$

Example with Units

$$0.5372 = \frac{0.21 \text{ mg/L}}{10^{\log_{10}(0.0003) + (0.23 \text{ d}^{-1} \cdot 0.5 \text{ d})}}$$








Evaluate Formula 



Variables used in list of Disposing of the Sewage Effluents Formulas above

- A_{DO} Actual Dissolved Oxygen (Milligram per Liter)
- C Mixing Concentration
- C_R River Concentration
- C_S Sewage Concentration
- d Depth of Stream (Meter)
- D Oxygen Deficit (Milligram per Liter)
- D_c Critical Oxygen Deficit
- D_o Initial Oxygen Deficit (Milligram per Liter)
- f Self-Purification Constant
- k Reoxygenation Coefficient per Sec (1 Per Second)
- K_D Deoxygenation Constant (1 Per Day)
- K_R Reoxygenation Coefficient (1 Per Day)
- $K_{R(20)}$ Reoxygenation Coefficient at Temperature 20 (1 Per Day)
- L Organic Matter at Start (Milligram per Liter)
- L_t Oxygen Equivalent (Milligram per Liter)
- Q_s Sewage Discharge (Cubic Meter per Second)
- Q_{stream} Discharge in Stream (Cubic Meter per Second)
- S_{DO} Saturated Dissolved Oxygen (Milligram per Liter)
- t Time in Days (Day)
- T Temperature (Kelvin)
- t_c Critical Time (Day)
- v Velocity (Meter per Second)

Constants, Functions, Measurements used in list of Disposing of the Sewage Effluents Formulas above


- **Functions:** \log , $\log(\text{Base}, \text{Number})$
Logarithmic function is an inverse function to exponentiation.
- **Functions:** \log_{10} , $\log_{10}(\text{Number})$
The common logarithm, also known as the base-10 logarithm or the decimal logarithm, is a mathematical function that is the inverse of the exponential function.
- **Functions:** sqrt , $\text{sqrt}(\text{Number})$
A square root function is a function that takes a non-negative number as an input and returns the square root of the given input number.
- **Measurement: Length** in Meter (m)
Length Unit Conversion 
- **Measurement: Time** in Day (d)
Time Unit Conversion 
- **Measurement: Temperature** in Kelvin (K)
Temperature Unit Conversion 
- **Measurement: Speed** in Meter per Second (m/s)
Speed Unit Conversion 
- **Measurement: Volumetric Flow Rate** in Cubic Meter per Second (m^3/s)
Volumetric Flow Rate Unit Conversion 
- **Measurement: Density** in Milligram per Liter (mg/L)
Density Unit Conversion 
- **Measurement: First Order Reaction Rate Constant** in 1 Per Day (d^{-1}), 1 Per Second (s^{-1})
First Order Reaction Rate Constant Unit Conversion 



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