

# Important Formulas on Surface Tension PDF



**Formulas  
Examples  
with Units**

## List of 17 Important Formulas on Surface Tension

### 1) Force given Surface Tension using Wilhelmy-Plate Method Formula

Formula

$$F = \left( \rho_p \cdot [g] \cdot (L \cdot B \cdot t) \right) + \left( 2 \cdot \gamma \cdot (t + B) \cdot (\cos(\theta)) \right) - \left( \rho_{\text{fluid}} \cdot [g] \cdot t \cdot B \cdot h_p \right)$$

Evaluate Formula

Example with Units

$$4.2E+9 \text{ N} = \left( 12.2 \text{ kg/m}^3 \cdot 9.8066 \text{ m/s}^2 \cdot (50 \text{ mm} \cdot 200 \text{ mm} \cdot 5000 \text{ mm}) \right) + \left( 2 \cdot 73 \text{ mN/m} \cdot (5000 \text{ mm} + 200 \text{ mm}) \cdot (\cos(15.1^\circ)) \right) - \left( 14.9 \text{ kg/m}^3 \cdot 9.8066 \text{ m/s}^2 \cdot 5000 \text{ mm} \cdot 20 \text{ mm} \right)$$

### 2) Height of Magnitude of Capillary Rise Formula

Formula

$$h_c = \frac{\gamma}{\left(\frac{1}{2}\right) \cdot (R \cdot \rho_{\text{fluid}} \cdot [g])}$$

Example with Units

$$12.1852 \text{ mm} = \frac{73 \text{ mN/m}}{\left(\frac{1}{2}\right) \cdot (82 \text{ mm} \cdot 14.9 \text{ kg/m}^3 \cdot 9.8066 \text{ m/s}^2)}$$

Evaluate Formula

### 3) Parachor given Surface Tension Formula

Formula

$$P_s = \left( \frac{M_{\text{molar}}}{\rho_{\text{liq}} \cdot \rho_v} \right) \cdot (\gamma)^{\frac{1}{4}}$$

Example with Units

$$2E-5 \text{ mol}^3/\text{mol}^3(\text{g/m}^3)^{3/4} = \left( \frac{44.01 \text{ g/mol}}{1141 \text{ kg/m}^3 \cdot 0.5 \text{ kg/m}^3} \right) \cdot (73 \text{ mN/m})^{\frac{1}{4}}$$

Evaluate Formula

### 4) Surface Pressure Formula

Formula

$$\Pi = \gamma_0 - \gamma$$

Example with Units

$$0.001 \text{ Pa} = 74 \text{ mN/m} - 73 \text{ mN/m}$$

Evaluate Formula

### 5) Surface Pressure using Wilhelmy-Plate Method Formula

Formula

$$\Pi = - \left( \frac{\Delta F}{2 \cdot (t + W_{\text{plate}})} \right)$$

Example with Units

$$0.0015 \text{ Pa} = - \left( \frac{-0.015 \text{ N}}{2 \cdot (5000 \text{ mm} + 16.9 \text{ g})} \right)$$

Evaluate Formula

### 6) Surface Tension for very Thin Plate using Wilhelmy-Plate Method Formula

Formula

$$\gamma = \frac{F_{\text{thin plate}}}{2 \cdot W_{\text{plate}}}$$

Example with Units

$$73.9645 \text{ mN/m} = \frac{0.0025 \text{ N}}{2 \cdot 16.9 \text{ g}}$$

Evaluate Formula

### 7) Surface Tension Force given Density of Fluid Formula

Formula

$$\gamma = \left(\frac{1}{2}\right) \cdot (R \cdot \rho_{\text{fluid}} \cdot [g] \cdot h_c)$$

Example with Units

$$59.9088 \text{ mN/m} = \left(\frac{1}{2}\right) \cdot (82 \text{ mm} \cdot 14.9 \text{ kg/m}^3 \cdot 9.8066 \text{ m/s}^2 \cdot 10 \text{ mm})$$

Evaluate Formula

### 8) Surface Tension given Contact Angle Formula

Formula

$$\gamma = \left( 2 \cdot R_{\text{curvature}} \cdot \rho_{\text{fluid}} \cdot [g] \cdot h_c \right) \cdot \left( \frac{1}{\cos(\theta)} \right)$$

Example with Units

$$75.6723 \text{ mN/m} = \left( 2 \cdot 25 \text{ mm} \cdot 14.9 \text{ kg/m}^3 \cdot 9.8066 \text{ m/s}^2 \cdot 10 \text{ mm} \right) \cdot \left( \frac{1}{\cos(15.1^\circ)} \right)$$

Evaluate Formula

### 9) Surface Tension given Correction Factor Formula

Formula

$$\gamma = \frac{m \cdot [g]}{2 \cdot \pi \cdot r_{\text{cap}} \cdot f}$$

Example with Units

$$75.3316 \text{ mN/m} = \frac{0.8 \text{ g} \cdot 9.8066 \text{ m/s}^2}{2 \cdot 3.1416 \cdot 32.5 \text{ mm} \cdot 0.51}$$

Evaluate Formula



### 10) Surface Tension given Critical Temperature Formula

Formula

$$\gamma_{Tc} = k_o \cdot \left( 1 - \left( \frac{T}{T_c} \right) \right)^{k_1}$$

Example with Units

$$39487.2323 \text{ mN/m} = 55 \cdot \left( 1 - \left( \frac{45 \text{ K}}{190.55 \text{ K}} \right) \right)^{1.23}$$

Evaluate Formula 

### 11) Surface Tension given Molar Volume Formula

Formula

$$\gamma_{MV} = [\text{EOTVOS\_C}] \cdot \frac{T_c - T}{\left( V_m \right)^{\frac{2}{3}}}$$

Example with Units

$$0.0038 \text{ mN/m} = 2.1\text{E-}7 \cdot \frac{190.55 \text{ K} - 45 \text{ K}}{\left( 22.4 \text{ m}^3/\text{mol} \right)^{\frac{2}{3}}}$$

Evaluate Formula 

### 12) Surface Tension given Molecular Weight Formula

Formula

$$\gamma = [\text{EOTVOS\_C}] \cdot \frac{T_c - T - 6}{\left( \frac{MW}{\rho_{\text{liq}}} \right)^{\frac{2}{3}}}$$

Example with Units

$$50.3956 \text{ mN/m} = 2.1\text{E-}7 \cdot \frac{190.55 \text{ K} - 45 \text{ K} - 6}{\left( \frac{16 \text{ g}}{1141 \text{ kg/m}^3} \right)^{\frac{2}{3}}}$$

Evaluate Formula 

### 13) Surface Tension given Temperature Formula

Formula

$$\gamma_T = 75.69 - (0.1413 \cdot T) - (0.0002985 \cdot (T)^2)$$

Example with Units

$$92389.9469 \text{ mN/m} = 75.69 - (0.1413 \cdot 45 \text{ K}) - (0.0002985 \cdot (45 \text{ K})^2)$$

Evaluate Formula 

### 14) Surface Tension of Pure Water Formula

Formula

$$\gamma_w = 235.8 \cdot \left( 1 - \left( \frac{T}{T_c} \right) \right)^{1.256} \cdot \left( 1 - \left( 0.625 \cdot \left( 1 - \left( \frac{T}{T_c} \right) \right) \right) \right)$$

Example with Units

$$87854.6012 \text{ mN/m} = 235.8 \cdot \left( 1 - \left( \frac{45 \text{ K}}{190.55 \text{ K}} \right) \right)^{1.256} \cdot \left( 1 - \left( 0.625 \cdot \left( 1 - \left( \frac{45 \text{ K}}{190.55 \text{ K}} \right) \right) \right) \right)$$

Evaluate Formula 

### 15) Total Weight of Plate using Wilhelmy-Plate Method Formula

Formula

$$W_{\text{tot}} = W_{\text{plate}} + \gamma \cdot (P) - U_{\text{drift}}$$

Example with Units

$$0.0202 \text{ N} = 16.9 \text{ g} + 73 \text{ mN/m} \cdot (250 \text{ mm}) - 15 \text{ mN/m}$$

Evaluate Formula 

### 16) Total Weight of Ring using Ring-Detachment Method Formula

Formula

$$W_{\text{tot}} = W_{\text{ring}} + (4 \cdot \pi \cdot r_{\text{ring}} \cdot \gamma)$$

Example with Units

$$0.0511 \text{ N} = 5 \text{ g} + (4 \cdot 3.1416 \cdot 0.502 \text{ mm} \cdot 73 \text{ mN/m})$$

Evaluate Formula 

### 17) Work of Cohesion given Surface Tension Formula

Formula

$$W_{\text{Coh}} = 2 \cdot \gamma \cdot [\text{Avaga-no}]^{\frac{1}{3}} \cdot \left( V_m \right)^{\frac{2}{3}}$$

Example with Units

$$9.8\text{E}+7 \text{ J/m}^2 = 2 \cdot 73 \text{ mN/m} \cdot 6\text{E}+23^{\frac{1}{3}} \cdot \left( 22.4 \text{ m}^3/\text{mol} \right)^{\frac{2}{3}}$$













Evaluate Formula 



## Variables used in list of Important Formulas on Surface Tension above

- **B** Width of Full Size Bearing Plate (Millimeter)
- **f** Correction Factor
- **F** Force (Newton)
- **F<sub>thin plate</sub>** Force on very Thin Plate (Newton)
- **h<sub>c</sub>** Height of Capillary Rise/Fall (Millimeter)
- **h<sub>p</sub>** Depth of Plate (Millimeter)
- **k<sub>1</sub>** Empirical Factor
- **k<sub>o</sub>** Constant for each Liquid
- **L** Length of Plate (Millimeter)
- **m** Drop Weight (Gram)
- **M<sub>molar</sub>** Molar Mass (Gram Per Mole)
- **MW** Molecular Weight (Gram)
- **P** Perimeter (Millimeter)
- **P<sub>s</sub>** Parachor (Cubic Meter per Mole (Joule per Square Meter)<sup>(0.25)</sup>)
- **R** Radius of Tubing (Millimeter)
- **r<sub>cap</sub>** Capillary Radius (Millimeter)
- **R<sub>curvature</sub>** Radius of Curvature (Millimeter)
- **r<sub>ring</sub>** Radius of Ring (Millimeter)
- **t** Thickness of Plate (Millimeter)
- **T** Temperature (Kelvin)
- **T<sub>c</sub>** Critical Temperature (Kelvin)
- **U<sub>drift</sub>** Upward Drift (Millinewton per Meter)
- **V<sub>m</sub>** Molar Volume (Cubic Meter per Mole)
- **W<sub>coh</sub>** Work of Cohesion (Joule per Square Meter)
- **W<sub>plate</sub>** Weight of Plate (Gram)
- **W<sub>ring</sub>** Weight of Ring (Gram)
- **W<sub>tot</sub>** Total Weight of Solid Surface (Newton)
- **Y** Surface Tension of Fluid (Millinewton per Meter)
- **Y<sub>MV</sub>** Surface Tension of Fluid given Molar Volume (Millinewton per Meter)
- **Y<sub>o</sub>** Surface Tension of Clean Water Surface (Millinewton per Meter)
- **Y<sub>T</sub>** Surface Tension of Fluid given Temperature (Millinewton per Meter)
- **Y<sub>Tc</sub>** Surface Tension of Fluid given Critical Temp (Millinewton per Meter)
- **Y<sub>w</sub>** Surface Tension of Pure Water (Millinewton per Meter)
- **ΔF** Change in Force (Newton)
- **θ** Contact Angle (Degree)
- **Π** Surface Pressure of Thin Film (Pascal)
- **ρ<sub>fluid</sub>** Density of Fluid (Kilogram per Cubic Meter)
- **ρ<sub>liq</sub>** Density of Liquid (Kilogram per Cubic Meter)
- **ρ<sub>p</sub>** Density of Plate (Kilogram per Cubic Meter)
- **ρ<sub>v</sub>** Density of Vapor (Kilogram per Cubic Meter)

## Constants, Functions, Measurements used in list of Important Formulas on Surface Tension above



- **constant(s): pi**, 3.14159265358979323846264338327950288  
Archimedes' constant
- **constant(s): [Avaga-no]**, 6.02214076E+23  
Avogadro's number
- **constant(s): [EOTVOS\_C]**, 0.00000021  
Eotvos constant
- **constant(s): [g]**, 9.80665  
Gravitational acceleration on Earth
- **Functions: cos**, cos(Angle)  
Cosine of an angle is the ratio of the side adjacent to the angle to the hypotenuse of the triangle.
- **Measurement: Length** in Millimeter (mm)  
Length Unit Conversion 
- **Measurement: Weight** in Gram (g)  
Weight Unit Conversion 
- **Measurement: Temperature** in Kelvin (K)  
Temperature Unit Conversion 
- **Measurement: Pressure** in Pascal (Pa)  
Pressure Unit Conversion 
- **Measurement: Force** in Newton (N)  
Force Unit Conversion 
- **Measurement: Angle** in Degree (°)  
Angle Unit Conversion 
- **Measurement: Heat Density** in Joule per Square Meter (J/m<sup>2</sup>)  
Heat Density Unit Conversion 
- **Measurement: Surface Tension** in Millinewton per Meter (mN/m)  
Surface Tension Unit Conversion 
- **Measurement: Density** in Kilogram per Cubic Meter (kg/m<sup>3</sup>)  
Density Unit Conversion 
- **Measurement: Molar Mass** in Gram Per Mole (g/mol)  
Molar Mass Unit Conversion 
- **Measurement: Molar Magnetic Susceptibility** in Cubic Meter per Mole (m<sup>3</sup>/mol)  
Molar Magnetic Susceptibility Unit Conversion 
- **Measurement: Parachor** in Cubic Meter per Mole (Joule per Square Meter)<sup>(0.25)</sup> (m<sup>3</sup>/mol<sup>(1/4)</sup>\*(J/m<sup>2</sup>)<sup>(1/4)</sup>)  
Parachor Unit Conversion 



## Download other Important Surface Chemistry PDFs

- [Important Freundlich adsorption isotherm Formulas](#) 

## Try our Unique Visual Calculators

-  Percentage increase 
-  HCF calculator 
-  Mixed fraction 

Please SHARE this PDF with someone who needs it!

## This PDF can be downloaded in these languages

[English](#) [Spanish](#) [French](#) [German](#) [Russian](#) [Italian](#) [Portuguese](#) [Polish](#) [Dutch](#)

7/10/2024 | 3:43:30 AM UTC

