

# Important Formulas in Gas Absorption & Stripping PDF

Formulas  
Examples  
with Units

List of 24  
Important Formulas in Gas Absorption &  
Stripping



## 1) Absorption Factor Formula ↗

Formula

$$A = \frac{L_s}{\alpha \cdot G_s}$$

Example with Units

$$1.7037 = \frac{23 \text{ mol/s}}{1.5 \cdot 9 \text{ mol/s}}$$

Evaluate Formula ↗

## 2) Absorption Factor given Stripping Factor Formula ↗

Formula

$$A = \frac{1}{S}$$

Example

$$0.7143 = \frac{1}{1.4}$$

Evaluate Formula ↗

## 3) Corrected Murphree Efficiency Percentage for Liquid Entrainment Formula ↗

Formula

$$E_{MGE} = \left( \frac{\frac{E_{MG}}{100}}{1 + \left( \left( \frac{E_{MG}}{100} \right) \cdot \left( \frac{E}{1 - E} \right) \right)} \right) \cdot 100$$

Evaluate Formula ↗

Example

$$55.914 = \left( \frac{\frac{65}{100}}{1 + \left( \left( \frac{65}{100} \right) \cdot \left( \frac{0.2}{1 - 0.2} \right) \right)} \right) \cdot 100$$

## 4) Gas Flowrate for Absorption Column on Solute Free Basis Formula ↗

Formula

$$G_s = \frac{L_s}{\frac{Y_{N+1} - Y_1}{X_N - X_0}}$$

Example with Units

$$9.5319 \text{ mol/s} = \frac{23 \text{ mol/s}}{0.8 - 0.1}$$

Evaluate Formula ↗



## 5) Gas Flowrate on Solute Free Basis for Inlet Conditions by Mole Fraction Formula

**Formula**

$$G_s = G_{N+1} \cdot (1 - y_{N+1})$$

**Example with Units**

$$18.9 \text{ mol/s} = 27 \text{ mol/s} \cdot (1 - 0.3)$$

**Evaluate Formula **

## 6) Gas Flowrate on Solute Free Basis for Inlet Conditions by Solute Free Mole Fraction Formula

**Formula**

$$G_s = \frac{G_{N+1}}{1 + Y_{N+1}}$$

**Example with Units**

$$15 \text{ mol/s} = \frac{27 \text{ mol/s}}{1 + 0.8}$$

**Evaluate Formula **

## 7) Liquid Flowrate for Absorption Column on Solute Free basis Formula

**Formula**

$$L_s = G_s \cdot \frac{Y_{N+1} - Y_1}{X_N - X_0}$$

**Example with Units**

$$21.7166 \text{ mol/s} = 9 \text{ mol/s} \cdot \frac{0.8 - 0.1}{0.3 - 0.0099}$$

**Evaluate Formula **

## 8) Liquid Flowrate on Solute Free Basis for Inlet Conditions by Solute Free Mole Fraction Formula

**Formula**

$$L_s = \frac{L_0}{1 + X_0}$$

**Example with Units**

$$24.7549 \text{ mol/s} = \frac{25 \text{ mol/s}}{1 + 0.0099}$$

**Evaluate Formula **

## 9) Liquid Flowrate on Solute Free Basis for Inlet Conditions using Mole Fraction Formula

**Formula**

$$L_s = L_0 \cdot (1 - x_1)$$

**Example with Units**

$$23.75 \text{ mol/s} = 25 \text{ mol/s} \cdot (1 - 0.05)$$

**Evaluate Formula **

## 10) Maximum Gas Rate for Absorption Column Formula

**Formula**

$$G_{smax} = \frac{L_s}{\frac{Y_{N+1} - Y_1}{\left(\frac{Y_{N+1}}{\alpha}\right) - X_0}}$$

**Example with Units**

$$17.1985 \text{ mol/s} = \frac{23 \text{ mol/s}}{\left(\frac{0.8}{1.5}\right) - 0.0099}$$

**Evaluate Formula **

## 11) Minimum Liquid Rate for Absorption Column Formula

**Formula**

$$L_{smin} = G_s \cdot \frac{Y_{N+1} - Y_1}{\left(\frac{Y_{N+1}}{\alpha}\right) - X_0}$$

**Example with Units**

$$12.0359 \text{ mol/s} = 9 \text{ mol/s} \cdot \left(\frac{0.8}{1.5}\right) - 0.0099$$

**Evaluate Formula **

## 12) Minimum Operating Line Slope for Absorption Column Formula

**Formula**

$$LsGs_{min} = \frac{Y_{N+1} - Y_1}{\left( \frac{Y_{N+1}}{\alpha} \right) - X_0}$$

**Example**

$$1.3373 = \frac{0.8 - 0.1}{\left( \frac{0.8}{1.5} \right) - 0.0099}$$

**Evaluate Formula **

## 13) Murphree Efficiency of Absorption Operation Based on Point Efficiency for Plug Flow Formula

**Formula**

$$E_{MG} = \left( A \cdot \left( \exp\left( \frac{E_{OG}}{A \cdot 100} \right) - 1 \right) \right) \cdot 100$$

**Evaluate Formula ****Example**

$$90.9983 = \left( 2 \cdot \left( \exp\left( \frac{75}{2 \cdot 100} \right) - 1 \right) \right) \cdot 100$$

## 14) Murphree Tray Efficiency of Absorption Operation Formula

**Formula**

$$E_{MG} = \left( \frac{y_n - y_{n+1}}{y_n^* - y_{n+1}} \right) \cdot 100$$

**Example**

$$53.5 = \left( \frac{0.557 - 0.45}{0.65 - 0.45} \right) \cdot 100$$

**Evaluate Formula **

## 15) Number of Absorption Stages by Kremser Equation Formula

**Formula**

$$N = \log_{10} \frac{\left( \frac{Y_{N+1} - (\alpha \cdot X_0)}{Y_1 - (\alpha \cdot X_0)} \right) \cdot \left( 1 - \left( \frac{1}{A} \right) \right) + \left( \frac{1}{A} \right)}{\log_{10}(A)}$$

**Evaluate Formula ****Example**

$$2.3534 = \log_{10} \frac{\left( \frac{0.8 - (1.5 \cdot 0.0099)}{0.1 - (1.5 \cdot 0.0099)} \right) \cdot \left( 1 - \left( \frac{1}{2} \right) \right) + \left( \frac{1}{2} \right)}{\log_{10}(2)}$$

## 16) Number of Stages for Absorption Factor Equal to 1 Formula

**Formula**

$$N = \frac{Y_{N+1} - Y_1}{Y_1 - (\alpha \cdot X_0)}$$

**Example**

$$8.2208 = \frac{0.8 - 0.1}{0.1 - (1.5 \cdot 0.0099)}$$

**Evaluate Formula **

## 17) Number of Stripping Stages by Kremser Equation Formula

[Evaluate Formula !\[\]\(21199eb166cc97331a0c54c649195dcc\_img.jpg\)](#)

**Formula**

$$N = \frac{\log_{10} \left( \left( \frac{x_0(\text{Stripping}) - \left( \frac{y_{N+1}(\text{Stripping})}{\alpha} \right)}{x_N(\text{Stripping}) - \left( \frac{y_{N+1}(\text{Stripping})}{\alpha} \right)} \right) \cdot \left( 1 - \left( \frac{1}{S} \right) \right) + \left( \frac{1}{S} \right) \right)}{\log_{10}(S)}$$

**Example**

$$6.0205 = \frac{\log_{10} \left( \left( \frac{0.225 - \left( \frac{0.001}{1.5} \right)}{0.01 - \left( \frac{0.001}{1.5} \right)} \right) \cdot \left( 1 - \left( \frac{1}{1.4} \right) \right) + \left( \frac{1}{1.4} \right) \right)}{\log_{10}(1.4)}$$

## 18) Operating Line Slope for Absorption Column Formula

[Evaluate Formula !\[\]\(23d9fc146e83b5c3013cfa32c784f8d5\_img.jpg\)](#)

**Formula**

$$LG_{\text{ratio}} = \frac{Y_{N+1} - Y_1}{X_N - X_0}$$

**Example**

$$2.413 = \frac{0.8 - 0.1}{0.3 - 0.0099}$$

## 19) Overall Tray Efficiency for Absorption Column given Murphree Efficiency Formula

[Evaluate Formula !\[\]\(aa53ad6fea213b8b2226d3077e30533a\_img.jpg\)](#)

**Formula**

$$E_O = \left( \frac{\ln \left( 1 + \left( \frac{E_{MG}}{100} \right) \cdot \left( \left( \frac{1}{A} \right) - 1 \right) \right)}{\ln \left( \frac{1}{A} \right)} \right) \cdot 100$$

**Example**

$$56.7041 = \left( \frac{\ln \left( 1 + \left( \frac{65}{100} \right) \cdot \left( \left( \frac{1}{2} \right) - 1 \right) \right)}{\ln \left( \frac{1}{2} \right)} \right) \cdot 100$$

## 20) Point Efficiency of Absorption Operation Formula

[Evaluate Formula !\[\]\(626ce8ac21792b9405bfddfea8e0c96a\_img.jpg\)](#)

**Formula**

$$E_{OG} = \left( \frac{y_{N, \text{Local}} - y_{N+1, \text{Local}}}{y_{\text{local, eqm}} - y_{N+1, \text{Local}}} \right) \cdot 100$$

**Example**

$$75 = \left( \frac{0.35 - 0.41}{0.33 - 0.41} \right) \cdot 100$$

## 21) Solute Free Mole Fraction of Gas in Inlet based on Mole Fraction Formula

[Evaluate Formula !\[\]\(c1168d6a8b365d11e842ece304635fa7\_img.jpg\)](#)

**Formula**

$$Y_{N+1} = \frac{y_{N+1}}{1 - y_{N+1}}$$

**Example**

$$0.4286 = \frac{0.3}{1 - 0.3}$$



## 22) Solute Free Mole Fraction of Liquid in Inlet based on Mole Fraction Formula

Formula

$$x_0 = \frac{x_1}{1 - x_1}$$

Example

$$0.0526 = \frac{0.05}{1 - 0.05}$$

Evaluate Formula 

## 23) Stripping Factor Formula

Formula

$$S = \frac{\alpha \cdot G_s(\text{Stripping})}{L_s(\text{Stripping})}$$

Example with Units

$$1.3948 = \frac{1.5 \cdot 25.2 \text{ mol/s}}{27.1 \text{ mol/s}}$$

Evaluate Formula 

## 24) Stripping Factor given Absorption Factor Formula

Formula

$$S = \frac{1}{A}$$

Example

$$0.5 = \frac{1}{2}$$

Evaluate Formula 



## Variables used in list of Important Formulas in Gas Absorption & Stripping above

- **A** Absorption Factor
- **E** Fractional Entrainment
- **E<sub>MG</sub>** Murphree Efficiency of Absorption Column
- **E<sub>MGE</sub>** Corrected Murphree Efficiency for Absorption
- **E<sub>O</sub>** Overall Tray Efficiency of Absorption Column
- **E<sub>OG</sub>** Point Efficiency of Absorption Column in Percent
- **G<sub>N+1</sub>** Inlet Gas Flowrate (Mole per Second)
- **G<sub>s</sub>** Gas Flowrate on Solute Free Basis (Mole per Second)
- **G<sub>s(Stripping)</sub>** Gas Flowrate on Solute Free Basis for Stripping (Mole per Second)
- **G<sub>smax</sub>** Maximum Gas Flowrate on Solute Free Basis (Mole per Second)
- **L<sub>0</sub>** Inlet Liquid Flowrate (Mole per Second)
- **L<sub>s</sub>** Liquid Flowrate on Solute Free Basis (Mole per Second)
- **L<sub>s(Stripping)</sub>** Liquid Flowrate on Solute Free Basis for Stripping (Mole per Second)
- **L<sub>smin</sub>** Minimum Liquid Flowrate on Solute Free Basis (Mole per Second)
- **LG<sub>ratio</sub>** Operating Line Slope of Absorption Column
- **LsGs<sub>min</sub>** Minimum Operating Line Slope of Absorption Column
- **N** Number of Stages
- **S** Stripping Factor
- **X<sub>0</sub>** Solute Free Mole Fraction of Liquid in Inlet
- **X<sub>0(Stripping)</sub>** Solute Free Mole Frac of Liquid in Stripping Inlet
- **X<sub>1</sub>** Liquid Inlet Mole Fraction
- **X<sub>N</sub>** Solute Free Mole Fraction of Liquid in Outlet

## Constants, Functions, Measurements used in list of Important Formulas in Gas Absorption & Stripping above

- **Functions:** **exp**, exp(Number)  
*n an exponential function, the value of the function changes by a constant factor for every unit change in the independent variable.*
- **Functions:** **ln**, ln(Number)  
*The natural logarithm, also known as the logarithm to the base e, is the inverse function of the natural exponential function.*
- **Functions:** **log10**, log10(Number)  
*The common logarithm, also known as the base-10 logarithm or the decimal logarithm, is a mathematical function that is the inverse of the exponential function.*
- **Measurement:** **Molar Flow Rate** in Mole per Second (mol/s)  
*Molar Flow Rate Unit Conversion* 



- $X_{N(\text{Stripping})}$  Solute Free Mole Frac of Liquid in Stripping Out
- $Y_1$  Solute Free Mole Fraction of Gas in Outlet
- $y_{\text{local, eqm}}$  Local Eqm Mole Fraction of Vapor on Nth Plate
- $y_n$  Average Mole Fraction of Vapour on Nth Plate
- $y_N, \text{Local}$  Local Mole Fraction of Vapor Leaving Nth Plate
- $y_{n+1}$  Average Mole Fraction of Vapour at N+1 Plate
- $y_{N+1}$  Gas Inlet Mole Fraction
- $Y_{N+1}$  Solute Free Mole Fraction of Gas in Inlet
- $Y_{N+1(\text{Stripping})}$  Solute Free Mole Frac of Gas in Stripping Inlet
- $y_{N+1, \text{Local}}$  Local Mole Fraction of Vapor Entering Nth Plate
- $y_n^*$  Average Mole Fraction at Equilibrium on Nth Plate
- $\alpha$  Equilibrium Constant for Mass Transfer

- [Important Gas Absorption Formulas](#) ↗

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